

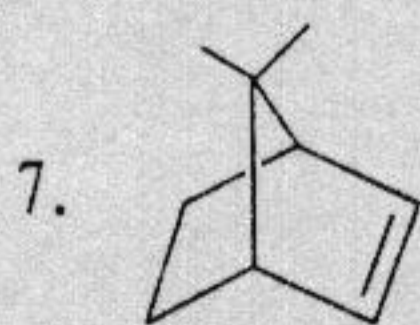
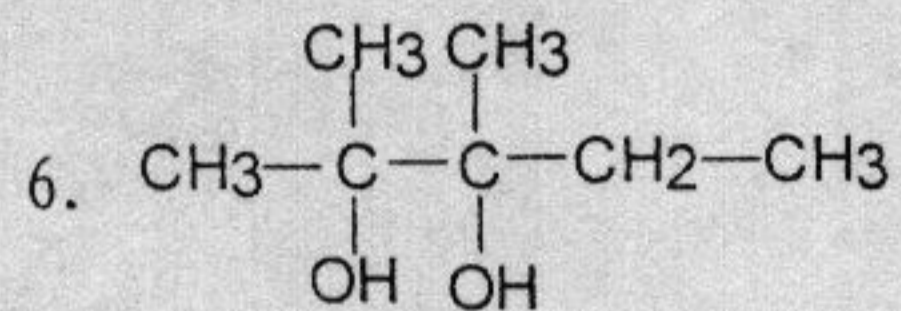
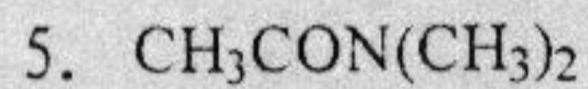
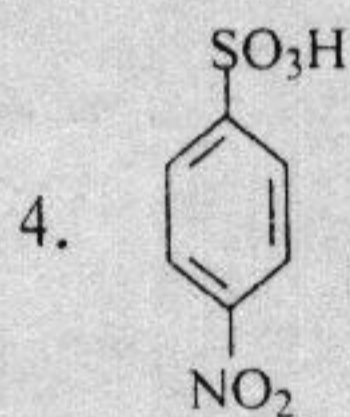
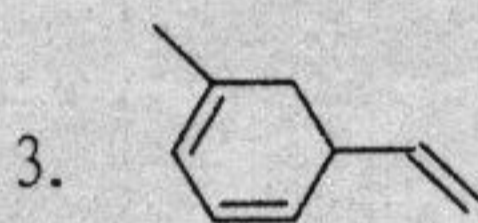
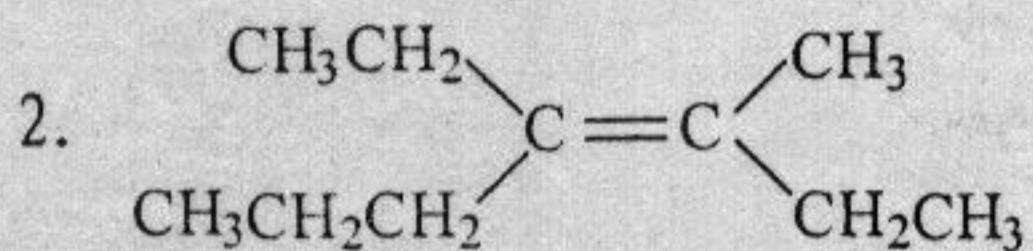
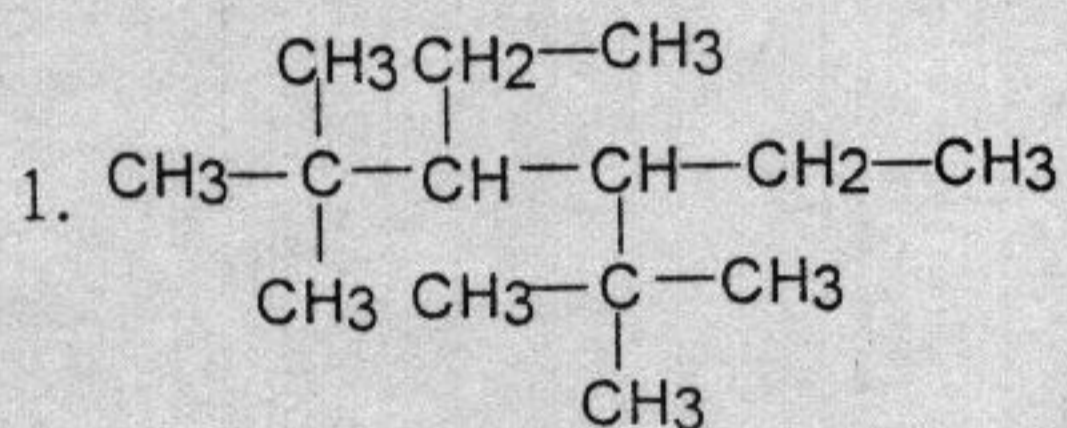
南京航空航天大学

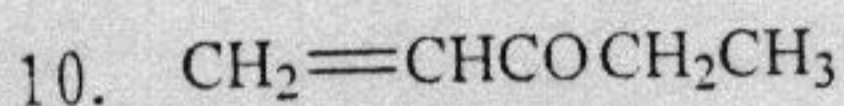
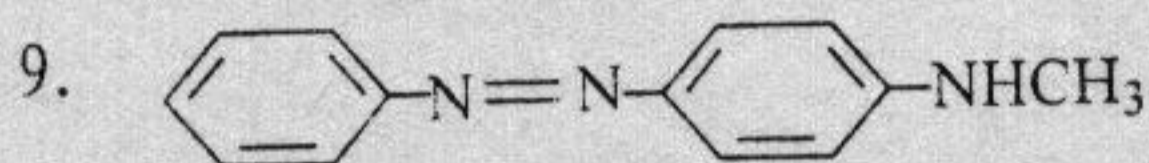
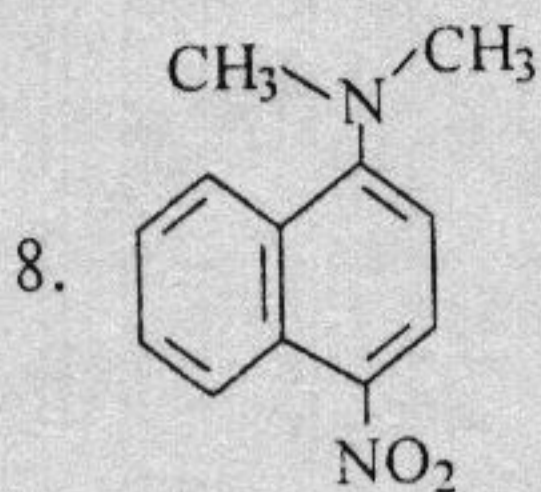
二〇〇四年硕士研究生入学考试试题

考试科目：有机化学

说明：答案一律写在答题纸上，写在试卷上无效

一、命名下列化合物（每题 1.5 分共 15 分）

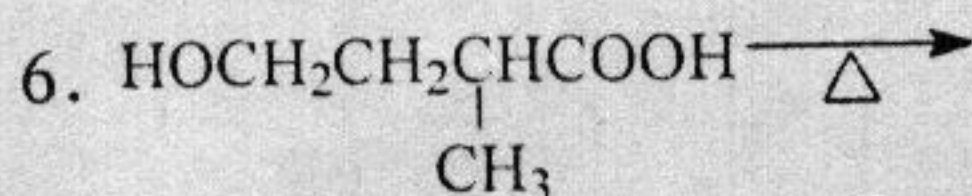
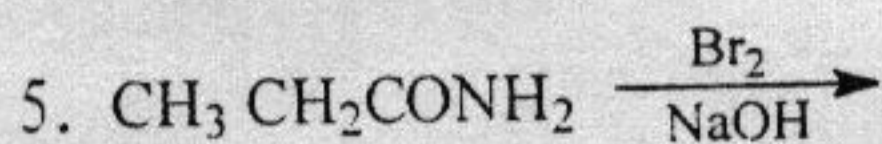
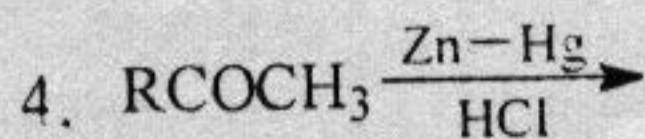
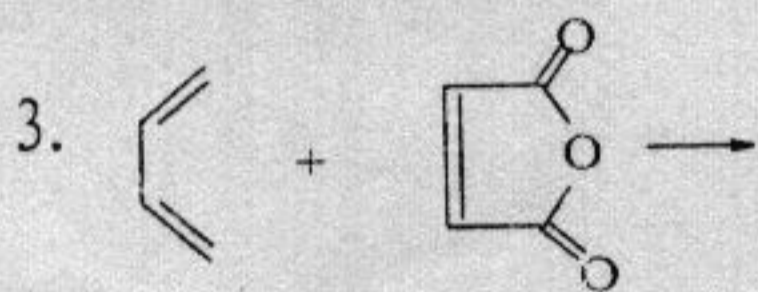
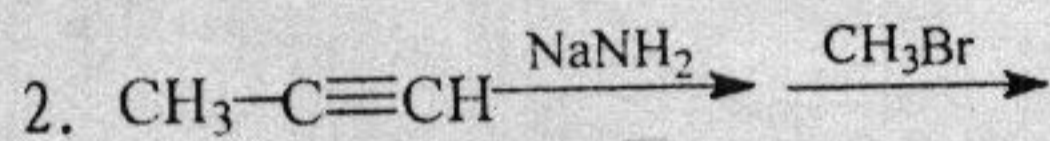
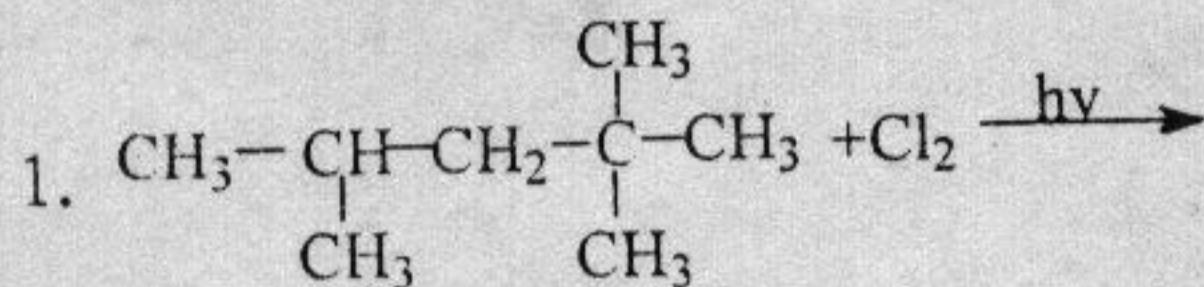


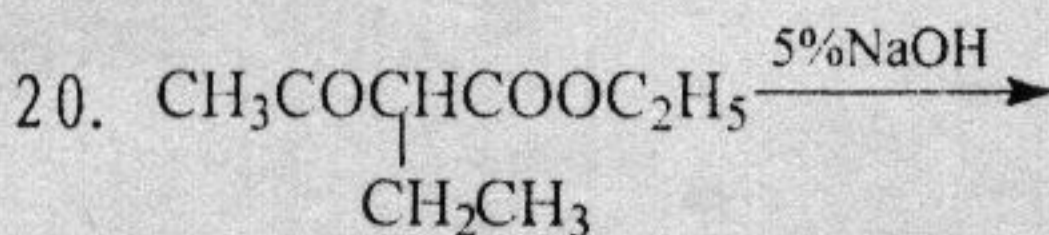
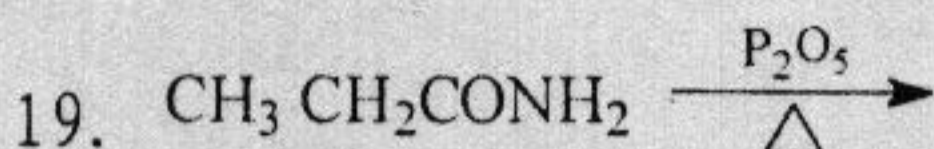
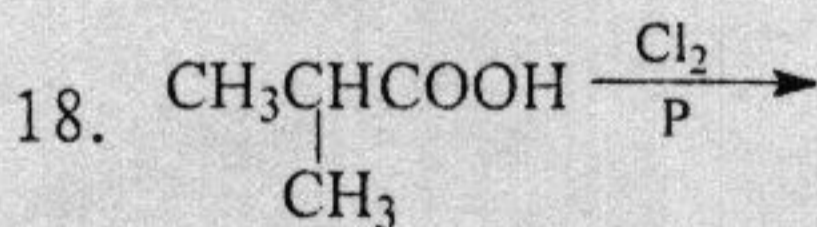
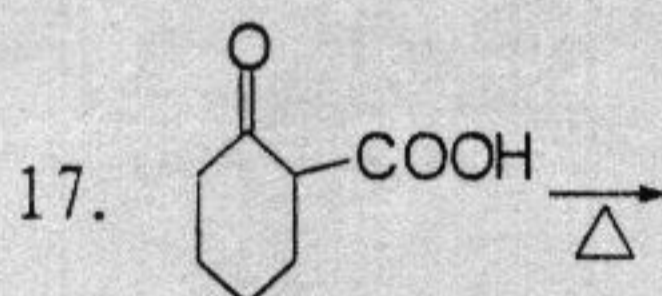
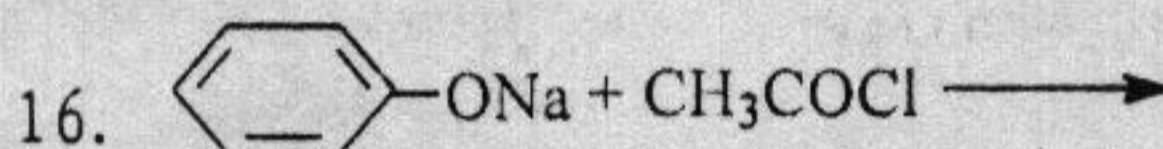
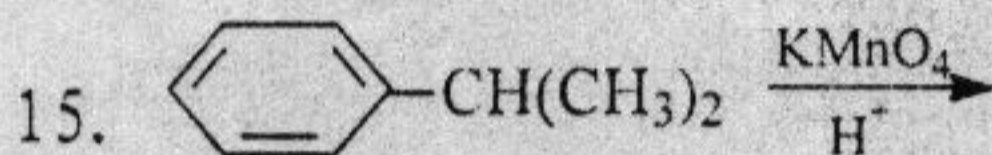
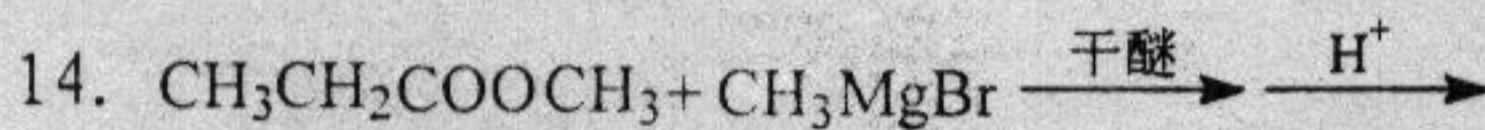
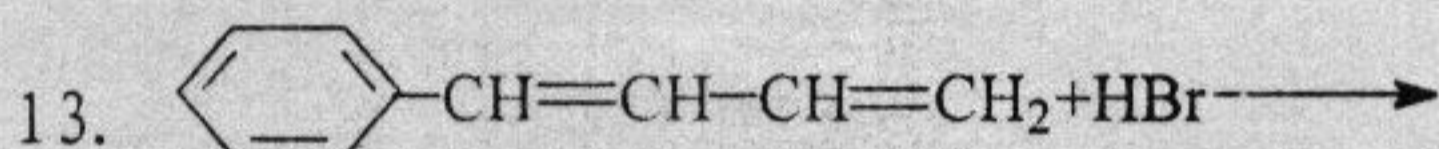
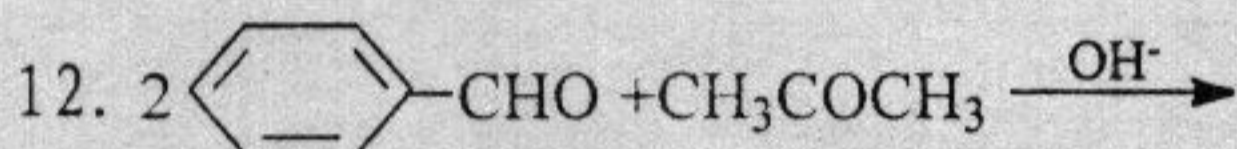
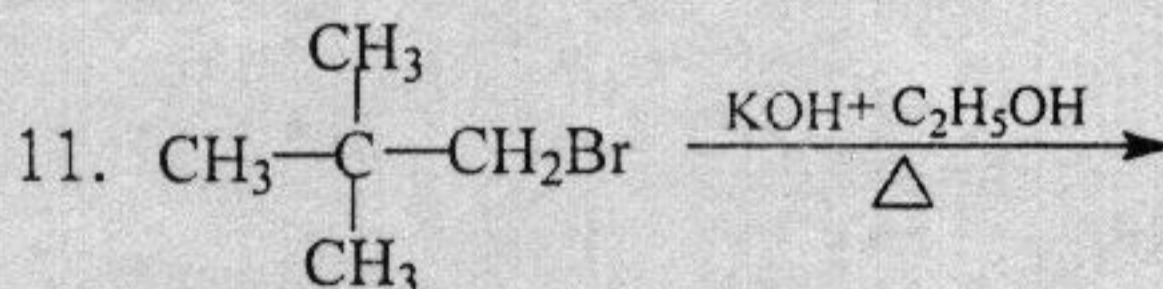
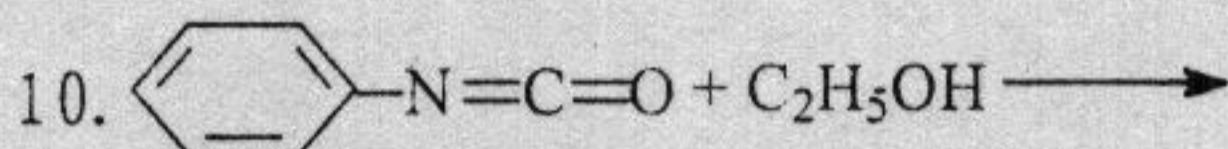
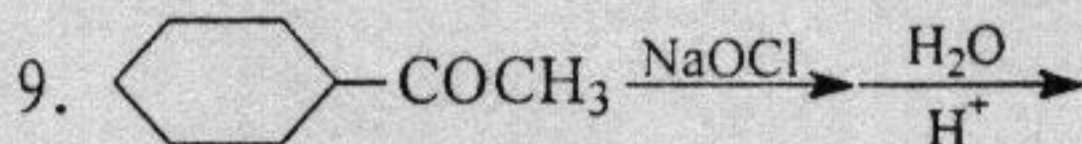
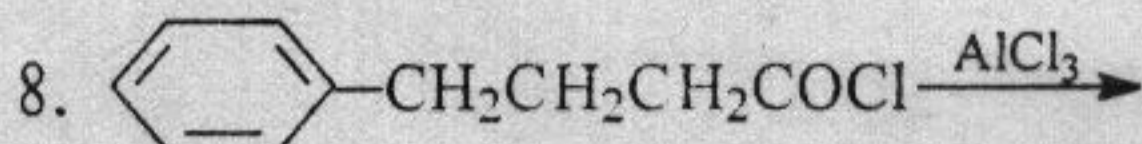
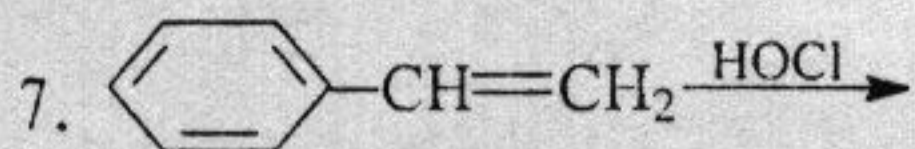


二、写出结构式 (每题 1.5 分共 15 分)

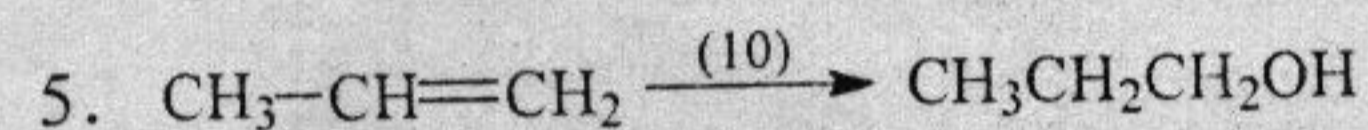
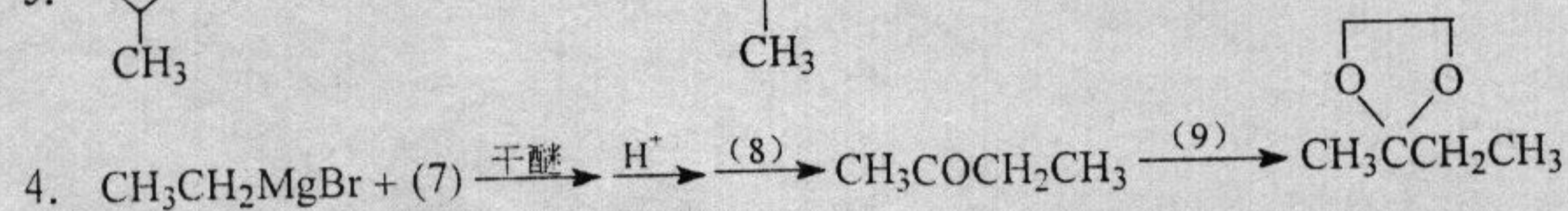
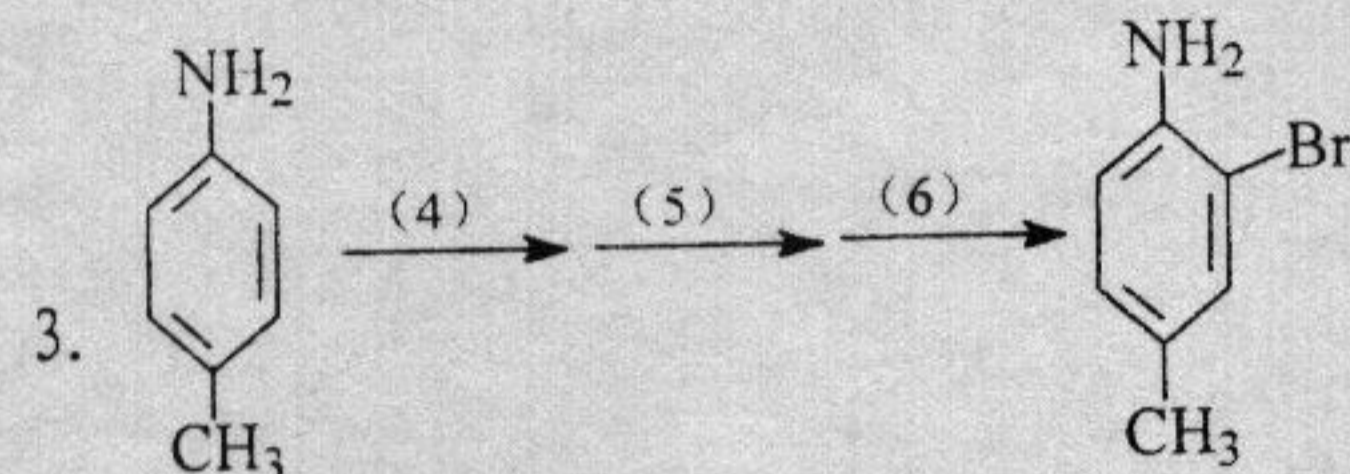
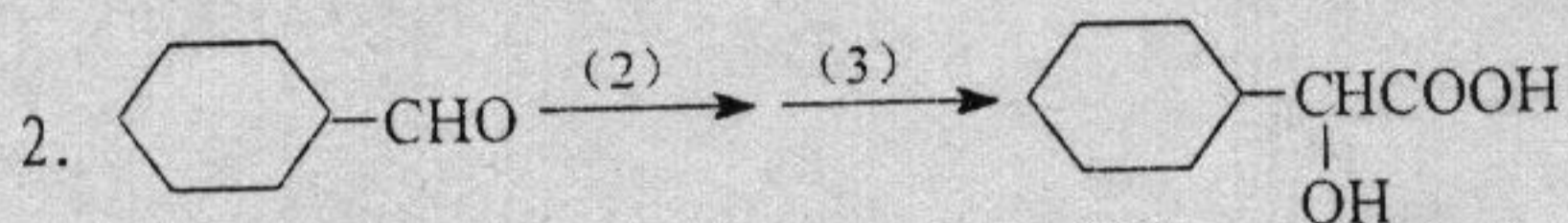
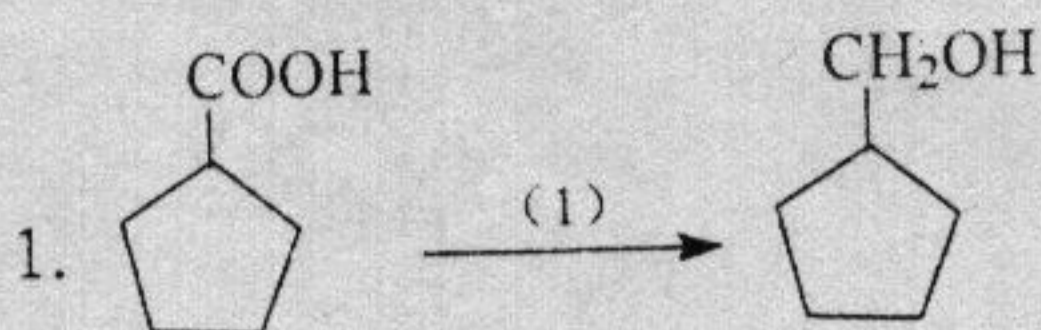
1. 异戊二烯
2. 螺[2,2]戊烷
3. 季戊四醇
4. 双酚 A (2, 2-(4, 4'-二羟基二苯基)丙烷)
5. 草酸
6. 邻苯二甲酰亚胺
7. ϵ -己内酰胺
8. 光气
9. 四氢呋喃
10. 重氮甲烷

三、写出下列反应的主要产物 (每题 2 分共 40 分)



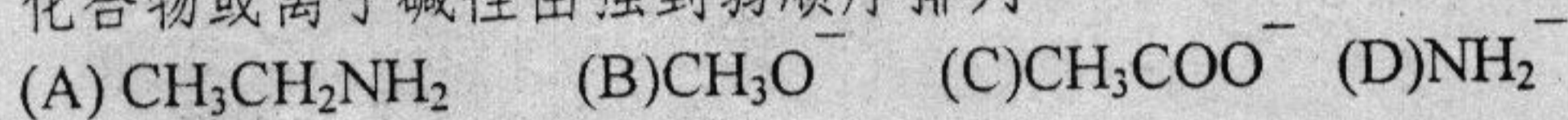


四、写出下列反应所需的试剂(每空 1 分共 10 分)

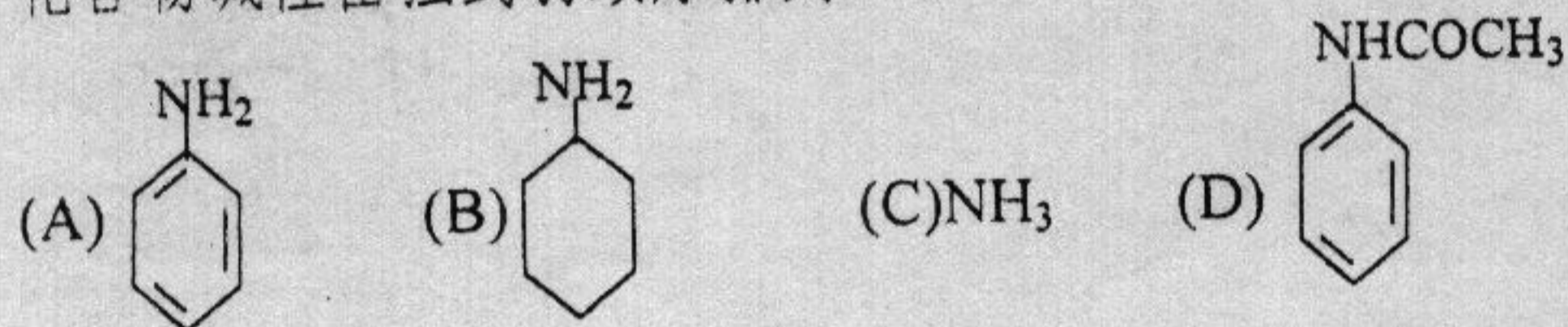


五、排序题(每题 2 分共 10 分)

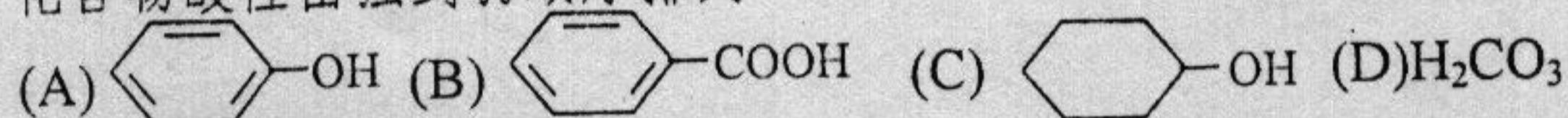
1. 化合物或离子碱性由强到弱顺序排列



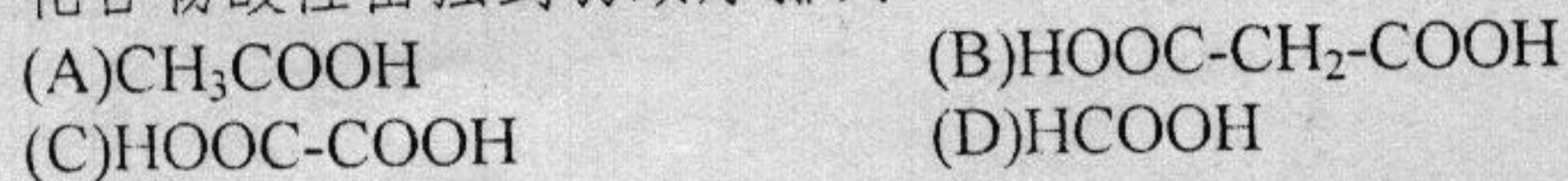
2. 化合物碱性由强到弱顺序排列



3. 化合物酸性由强到弱顺序排列



4. 化合物酸性由强到弱顺序排列



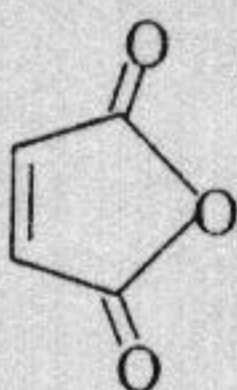
5. 化合物水解反应活性由强到弱顺序排列



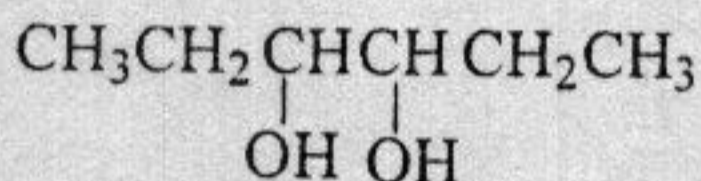
六、HCl 与 $\text{CH}_2=\text{CH}-\overset{\text{CH}_3}{\underset{\text{CH}_3}{\text{C}}}-\text{CH}_3$ 加成得到 $\text{CH}_3-\overset{\text{CH}_3}{\underset{\text{CH}_3}{\text{C}}}-\underset{\text{Cl}}{\text{CH}}-\text{CH}_3$, $\text{CH}_3-\overset{\text{Cl}}{\underset{\text{CH}_3}{\text{C}}}-\underset{\text{CH}_3}{\text{CH}}-\text{CH}_3$
 写出并解释每种产物形成过程。(10分)

七、合成题 (无机试剂任选) (每题 6 分共 36 分)

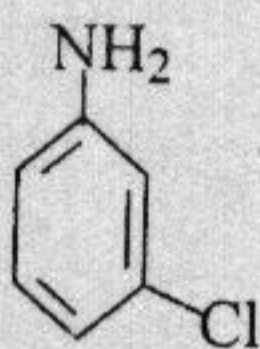
1. 乙醇为原料合成



2. 乙炔为原料合成



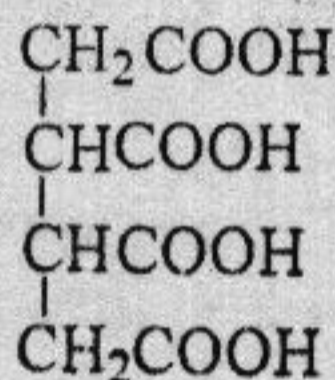
3. 甲苯为原料合成



4. 乙烯为原料合成 $\text{CH}_3\text{CH}_2\text{COCH}_3$

5. 丙稀为原料合成 2-甲基戊烷

6. 丙二酸二乙酯与合适的有机试剂合成



八、推断题 (每题 7 分共 14 分)

1. 某二元羧酸 A ($\text{C}_8\text{H}_{14}\text{O}_4$) 高温加热时转变为非酸化合物 B ($\text{C}_7\text{H}_{12}\text{O}$), B 用浓硝酸氧化时生成二元羧酸 C ($\text{C}_7\text{H}_{12}\text{O}_4$), 加热时 C 生成酸酐 D ($\text{C}_7\text{H}_{10}\text{O}_3$); 用 LiAlH_4 还原时, A 转化为 E ($\text{C}_8\text{H}_{18}\text{O}_2$), 该 E 能脱水生成 3,4-甲基-1,5-己二烯, 推断化合物 A、B、C、D、E 的结构。

2. 一未知酸 A ($\text{C}_9\text{H}_{10}\text{O}_3$) 与 $\text{CrO}_3-\text{H}_2\text{SO}_4$ 作用生成 B ($\text{C}_9\text{H}_8\text{O}_3$). B 可发生银镜反应生成 C ($\text{C}_9\text{H}_8\text{O}_4$), C 受热容易转变成 D ($\text{C}_8\text{H}_8\text{O}_2$). A 用 KMnO_4/H^+ 氧化得苯甲酸。推断化合物 A、B、C、D 的结构。